

Irrev, Pring 4

1a) $\vec{J} = L \vec{X}$, $\vec{J} = \begin{pmatrix} J_1 \\ J_2 \\ J_2' \end{pmatrix}$, $L = \begin{bmatrix} L_{11} & L_{12} & L_{12'} \\ L_{21} & L_{22} & L_{22'} \\ L_{2'1} & L_{2'2} & L_{2'2'} \end{bmatrix}$, $\vec{X} = \begin{pmatrix} -\frac{1}{T} \frac{\partial \mu_1}{\partial x} \\ -\frac{1}{T} \frac{\partial \mu_2}{\partial x} \\ \frac{\partial}{\partial x} \left(\frac{1}{T} \right) \end{pmatrix}$

1b) Mass transport due to a temperature gradient
 $\frac{\partial T}{\partial x} \neq 0 \xrightarrow{\text{Soret}} J_i \neq 0$

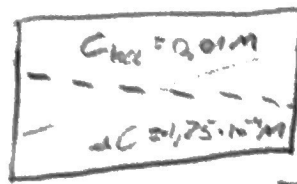
1c) For a one-component system: $D_T = \frac{L_{12}}{C_1 T^2}$
 $S_T := \frac{D_T}{D_{1,2}}$, $D_{1,2} = \frac{L_{11} R}{C_1} \left(1 + \frac{\partial \ln \gamma_1}{\partial \ln C_1} \right)$,
 $\alpha_T := S_T T$

1d) Heat transport due to a chemical potential gradient

1e) $q^* := \left(\frac{J_2'}{J_1} \right)_{dT=0} = \frac{L_{21}}{L_{11}} = \frac{C_1 D_T T^2}{D_{1,2} T} \frac{\partial \mu_{1,T}}{\partial C_1}$

1f) $D_{1,2}$ can be estimated by $D_{1,1} = D_S$, the self-diffusion coefficient, which can be measured by NMR or calculated for gases using kinetic gas theory.

Oppg 2)



$$D_{1,2} = 1,9 \cdot 10^{-9} \frac{\text{m}^2}{\text{s}}$$

$$T_1 = 293 \text{ K}$$

$$T_2 = 303 \text{ K}$$

Za) $S_T = \frac{D_T}{D_{1,2}} = \frac{-\frac{\partial \epsilon_1}{\partial x}}{C_1 \frac{\partial T}{\partial x}}$ steady state $\rightarrow S_T = \frac{-\Delta C_1}{C_1 \Delta T} = -1,25 \cdot 10^{-3} \text{ K}^{-1}$

$$q^* = S_T R T^2 \left(1 + \frac{\partial \ln \gamma_1}{\partial \ln C_1} \right)$$

$$= S_T R T^2 \left(1 + C_1 \frac{\partial \ln \gamma_1}{\partial C_1} \right)$$

$$= S_T R T^2 \left[1 + C_1 \frac{d}{dC_1} \left(-\sqrt{\frac{C_1}{M_{\text{OH}}}} \right) \right]$$

$$= S_T R T^2 \left[1 - \frac{C_1}{M_{\text{OH}}} \frac{1}{2} \right]$$

$$= S_T R T^2 \left(1 - \frac{1}{2} \sqrt{\frac{C_1}{M_{\text{OH}}}} \right)$$

$$= 0,94 \rightarrow \text{definitely ideal}$$

$$q^* = -867,5 \frac{\text{J}}{\text{mol}}$$

(Also: is q^* defined "per component"?)

$$d \ln C = \frac{1}{C} dC$$

$$\gamma_1 \approx \exp \left(z_1 z_2 \sqrt{\frac{C_1}{M_{\text{OH}}}} \right) = 0,89$$

$$\mu = \frac{1}{2} \sum z_i^2 C_i = C_1 = 0,01 \text{ mol}$$

can probably assume ideal

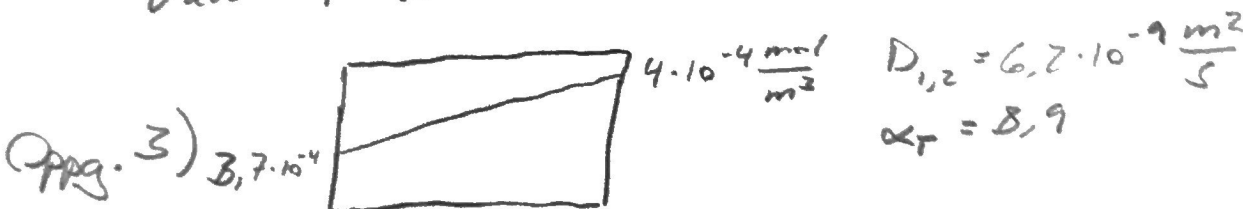
$$\frac{d \ln \gamma}{d \ln C} = \frac{z_1 z_2}{2} \sqrt{\frac{C_1}{M_{\text{OH}}}}$$

For $A, B, \dots, X, Y, (aq)$

Zb) $J_{HCl} = L_{H_2} \frac{\partial}{\partial x} \left(\frac{1}{T} \right)$, $L_{H_2} = D_T C_1 T^2$, $D_T = S_T D_{1,2}$
 $L_{H_2} = S_T D_{1,2} C_1 T^2$

$$J_{HCl} = \frac{-L_{H_2}}{T^2} \frac{\partial T}{\partial x}$$

$$J_{HCl} = -S_T D_{1,2} C_1 \frac{\Delta T}{\Delta x} = -2,25 \cdot 10^{-8} \frac{\text{mol}}{\text{m}^2 \text{s}}$$



$$S_T = \frac{\alpha_T}{T}, D_T = S_T D_{1,2} = \frac{\alpha_T D_{1,2}}{T}$$

$$J_2' = -\frac{L_{H_2}}{T} \frac{\partial \mu_{1,T}}{\partial x}, L_{H_2} = L_{H_2} = D_T C_1 T^2 = \alpha_T D_{1,2} C_1 T$$

$$\mu_1 = \mu_1^0 + RT \ln \gamma_1 \frac{P_1}{P^0}, \frac{P_1}{P^0} = \frac{C_1 RT}{C^0 RT} = \frac{C_1}{C^0}$$

$$\frac{\partial \mu_{1,T}}{\partial x} = RT \frac{\partial \ln \gamma_1}{\partial x}, \text{ assuming } \gamma = 1$$

$$= \frac{RT}{C_1} \frac{\partial C_1}{\partial x}$$

$$J_2' = -\alpha_T D_{1,2} RT \frac{\Delta C}{\Delta x}$$

$$J_2' = -4,76 \cdot 10^{-8} \frac{\text{J}}{\text{m}^2 \text{s}}$$