

Rekth, øving 9, Vegard G. Jørvell

Oppg. 1)

$$X_A = 0,7$$

$$\Gamma_A = \frac{F_{A0} - F_A}{V} \rightarrow V = \frac{F_{A0} X_A}{\Gamma_A} \quad (I)$$

$$II) \Gamma_A = k_1 C_A, \quad k_1 = \exp\left(17,2 - \frac{\bar{E}_1}{RT}\right), \quad \bar{E}_1 = 11600 \frac{\text{Cal}}{\text{mol}}$$

$$\Delta_{rx} H = -18000 \frac{\text{Cal}}{\text{mol}}$$

$$III) \Delta T = \frac{-X_A \Delta_{rx} H}{C_p}, \quad \text{når } \Delta_{rx} C_p = 0$$

$$C_p = 1000 \frac{\text{kcal}}{\text{m}^3 \text{K}} \cdot \frac{1}{3} \frac{\text{m}^3}{\text{kmol}} = \frac{1000}{3} \frac{\text{Cal}}{\text{mol} \cdot \text{K}}$$

$$\Delta T = 37,8 \text{ K} \rightarrow \underline{T_{rx} = 57,8^\circ \text{C} = 330,8 \text{ K}}$$

$$II) \underline{k_1 = 0,639 \text{ min}^{-1}}, \quad R = 1,987 \frac{\text{Cal}}{\text{mol} \cdot \text{K}}$$

$$\Gamma_A = k C_A = k C_{A0} (1 - X)$$

$$II) \underline{\Gamma_A = 0,575 \frac{\text{kmol}}{\text{m}^3 \text{min}}}$$

$$I) \underline{\underline{V = 7,3 \text{ m}^3}}$$

Oppg. 2)

Antar at katalysediet har konstant temp.

$$A = 100 \text{ m}^2 \quad U = 10 \frac{\text{kcal}}{\text{m}^2 \text{ K min}} \quad T_a = 10^\circ \text{C}$$

Energibalans:

$$Q + H_{\text{in}} - H_{\text{ut}} = 0$$

$$UA(T_a - T) + \sum F_{i0} \Delta T H_i - \sum F_i (\Delta T H_i + C_{p,i} (T - T_a)) = 0$$

$$UA(T_a - T) + \sum F_{A0} \theta_i \Delta T H_i - \sum F_{A0} (\theta_i + \nu_i X_A) (\Delta T H_i + C_{p,i} (T - T_0)) = 0$$

$$\frac{UA}{F_{A0}} (T_a - T) + X_A \Delta_{\text{rx}} H - \sum \theta_i C_{p,i} (T - T_0) - \underbrace{\Delta_{\text{rx}} C_p (T - T_0)}_{=0} = 0$$

$$X_A \Delta_{\text{rx}} H = \frac{UA}{F_{A0}} T_a - T \left(\frac{UA}{F_{A0}} + C_{p,A} \right) + T_0 C_{p,A}$$

$$T = \frac{\frac{UA}{F_{A0}} T_a + T_0 C_{p,A} - X_A \Delta_{\text{rx}} H}{\frac{UA}{F_{A0}} + C_{p,A}} \quad \left\{ \begin{array}{l} \theta_A = 1, \theta_R = 0 \end{array} \right.$$

$$\underline{T = 314,9 \text{ K}} \quad \rightarrow \underline{k_1 = 0,262}$$

$$\Gamma_A = \frac{X_A F_{A0}}{V_{\text{rx}}} = k_1 C_{A0} (1 - X_A)$$

$$X_A \left(\frac{F_{A0}}{V_{\text{rx}}} + k_1 C_{A0} \right) = k_1 C_{A0}$$

$$\underline{\underline{X_A = 0,489}}$$

Oppg. 3)

$$\left. \begin{aligned} r_A &= k_1 C_A - k_2 C_R \\ C_A &= C_{A0} (1 - x_A) \\ C_R &= x_A C_{A0} \end{aligned} \right\} r_A = C_{A0} (k_1 (1 - x_A) - k_2) \quad \textcircled{I}$$

$$\text{II) } k_1 = A_1 \exp\left(-\frac{\bar{E}_1}{RT}\right) \quad \text{III) } k_2 = A_2 \exp\left(-\frac{\bar{E}_2}{RT}\right)$$

$$\frac{dr_A}{dT} = C_{A0} \left(\frac{dk_1}{dT} (1 - x_A) - \frac{dk_2}{dT} \right)$$

$$\frac{dk_i}{dT} = \frac{A_i \bar{E}_i}{RT^2} \exp\left(-\frac{\bar{E}_i}{RT}\right)$$

$$\frac{dr_A}{dT} = 0 \rightarrow \frac{A_1 \bar{E}_1}{RT^2} \exp\left(-\frac{\bar{E}_1}{RT}\right) (1 - x_A) = \frac{A_2 \bar{E}_2}{RT^2} \exp\left(-\frac{\bar{E}_2}{RT}\right)$$

$$\exp\left(\frac{\bar{E}_2 - \bar{E}_1}{RT}\right) = \frac{A_2 \bar{E}_2}{A_1 \bar{E}_1 (1 - x_A)} = 3,7 \cdot 10^{11}$$

$$T = \frac{\bar{E}_2 - \bar{E}_1}{R \ln(3,7 \cdot 10^{11})}$$

$$T = 340 \text{ K}$$

$$\underline{\underline{r(340 \text{ K}) = 0,56 \frac{\text{kmol}}{\text{m}^3 \text{min}}}}$$